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# $\mathbf{C}-\mathbf{H} \cdots \mathbf{N}$ and $\mathbf{C}-\mathbf{H} \cdots \pi$ interactions in 2-ethoxy-4,6-diphenylpyridine-3-carbonitrile 

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The title compound, $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}$, has two molecules in the asymmetric unit and the crystal structure shows that the central pyridine ring of each molecule has a flat boat conformation. The terminal C atom in one of the molecules is disordered over two positions, with relative occupancies of 0.594 (14) and 0.398 (14). Intermolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{N}$ and $\mathrm{C}-$ $\mathrm{H} \cdots \pi$ interactions and $\pi-\pi$ stacking, along with intramolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{N}$ and $\mathrm{C}-\mathrm{H} \cdots \pi$ interactions, help to stabilize the structure.

## Comment

The title compound, (I), belongs to the 2-pyridine class of compounds, which show a wide spectrum of biological activities (Perez-Medina et al., 1947). The present work continues our structural studies of the 2-pyridine group of compounds (Patel et al., 2002a,b).

(I)

Compound (I) crystallizes in the monoclinic space group $P 2_{1} / c$ with two molecules in the asymmetric unit, denoted $A$ and $B$. A perspective view of the two molecules is shown in Fig. 1 (Johnson, 1965). The pyridine rings of molecules $A$ and

[^0]$B$ have quite similar geometries. The shortening of the $\mathrm{C}-\mathrm{N}$ distances $[1.313$ (3) and 1.346 (3) $\AA$ in molecule $A$, and 1.349 (3) and 1.299 (3) $\AA$ in molecule $B]$ and the opening of the $\mathrm{N} 1-\mathrm{C} 1-\mathrm{C} 2$ angle $\left[123.9(2)^{\circ}\right.$ in molecule $A$ and $124.6(3)^{\circ}$ in molecule $B$ ] may be attributed to the size of the substituent at C 1 , correlating well with the values observed in the ortho-substituted derivative 2-(2,3-isopropylidenedioxy-propyloxy)-3-cyano-4,6-diphenylpyridine (Patel et al., 2002c). However, when the ortho substituent is only an O atom, the bond length and angle are normal (Patel et al., 2002a,b). The bond distances and angles of the phenyl rings of molecules $A$ and $B$ in (I) are consistent with those observed in other similar structures (Patel et al., 2002b; Black et al., 1992; Hussain et al., 1996).

The flat boat conformation of the dihydropyridine ring of molecules $A$ and $B$ is reported in the literature for other compounds incorporating this moiety (Lokaj et al., 1991; Fonseca et al., 1986). The cyano atoms C18 and N2 of molecule $A$ deviate by 0.173 (4) and 0.310 (5) $\AA$, respectively, from the pyridine ring plane, while atoms $\mathrm{C} 18^{\prime}[-0.031(4) \AA]$ and $\mathrm{N}^{\prime}$ $[-0.058(6) \AA]$ are coplanar with the pyridine plane in molecule $B$. The phenyl rings of both molecules are planar. Steric hindrances rotate the phenyl rings out of the plane of the central pyridine ring by 35.49 (8) and 24.61 (1) ${ }^{\circ}$ (molecule $A$ ), and by 41.0 (1) and $22.3(1)^{\circ}$ (molecule $B$ ). A similar orientation of the phenyl rings has been observed in other derivatives of 1,4-dihydropyridine reported by our group and by others (Patel et al., 2002a; Bolte, 1998).

The significant difference in the conformations of molecules $A$ and $B$ is in the relative orientations of the ethoxy moiety. The ethoxy group in molecule $B$ is orthogonal to the pyridine ring plane $\left[\mathrm{C1}^{\prime}-\mathrm{O}^{\prime}-\mathrm{C} 19^{\prime}-\mathrm{C} 20^{\prime}=94.1(6)^{\circ}\right]$, thereby avoiding steric interaction, while in molecule $A$ it is coplanar with the heterocyclic ring plane $[\mathrm{C} 1-\mathrm{O} 1-\mathrm{C} 19-\mathrm{C} 20=$ $\left.170.1(2)^{\circ}\right]$.

The terminal ethoxy C atom of molecule $B$ is disordered over two positions, resulting in a shortening of the $\mathrm{C} 19^{\prime}-\mathrm{C} 20^{\prime}$ $\left[1.350(6) \AA\right.$ ] and $\mathrm{C} 19^{\prime}-\mathrm{C} 20 A$ [1.392 (9) A ] distances. The torsion angles defining the orientations of various substituents at the central pyridine ring are summarized as follows for molecule $A$, with the values for molecule $B$ given in brackets: $\mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 18-\mathrm{N} 2=146(6)^{\circ}\left[35(43)^{\circ}\right], \mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 18-\mathrm{N} 2=$ $-33(6)^{\circ} \quad[-149(43)], \quad \mathrm{C} 2-\mathrm{C} 1-\mathrm{O} 1-\mathrm{C} 19=-172.0(2)^{\circ}$ $\left[179.1(3)^{\circ}\right]$ and $\mathrm{N} 1-\mathrm{C} 1-\mathrm{O} 1-\mathrm{C} 19=8.1(3)^{\circ}\left[-0.6(4)^{\circ}\right]$

An intermolecular network of $\mathrm{C}-\mathrm{H} \cdots \pi$ and $\pi-\pi$-stacking interactions plays a significant role in the stability of the structure of (I). In addition, there is an intramolecular C$\mathrm{H} \cdots \mathrm{N}$ interaction (Fig. 2) involving pyridine atom $\mathrm{N} 1^{\prime}$ and ethoxy atom $\mathrm{C} 19^{\prime}\left[\mathrm{C} 19^{\prime} \cdots \mathrm{N} 1^{\prime}=2.709(5) \AA, \mathrm{H} 19 D \cdots \mathrm{~N} 1^{\prime}=\right.$ $2.34 \AA$ and $\mathrm{C}_{1} 9^{\prime}-\mathrm{H} 19 D \cdots \mathrm{~N} 1^{\prime}=102^{\circ}$ ]. Cyano atom N 2 of molecule $A$ interacts with phenyl atom $\mathrm{C}^{\prime} 5^{\prime}$ of a symmetryrelated molecule $B$ at $\left(x, \frac{3}{2}-y, z-\frac{1}{2}\right)\left[C 15^{\prime} \cdots \mathrm{N} 2=\right.$ 3.399 (6) $\AA, \mathrm{H} 15^{\prime} \cdots \mathrm{N} 2=2.55^{2} \AA$ and $\left.\mathrm{C} 15^{\prime}-\mathrm{H} 15^{\prime} \cdots \mathrm{N} 2=153^{\circ}\right]$. The hydrogen bonding in the structure of (I) is supported by $\pi-\pi$ stacking between the pyridine ring plane of molecule $A$ and that of molecule $B$ at $(1+x, y, z)$, with their centroids separated by 3.926 (2) $\AA$.


Figure 1
A view of the two independent molecules in (I), showing the atomnumbering scheme; molecule $B$ is labelled with primes. Displacement ellipsoids are drawn at the $30 \%$ probability level and H atoms have been omitted for clarity.

There are two intramolecular $\mathrm{C}-\mathrm{H} \cdots \pi$ interactions in the structure of (I). In the first, phenyl atom $\mathrm{C13}^{\prime}$ of molecule $B$ interacts with the pyridine ring of molecule $A\left[\mathrm{C13}^{\prime} \cdots C g 1=\right.$ $3.392(3) \AA, \mathrm{H} 13^{\prime} \cdots C g 1=3.18 \AA$ and $\mathrm{C13}^{\prime}-\mathrm{H} 13^{\prime} \cdots C g 1=$ $95^{\circ} ; C g 1$ is the centroid of the $\mathrm{C} 1-\mathrm{C} 5 / \mathrm{N} 1$ ring]. In the second interaction, ethoxy atom C 19 of molecule $A$ interacts with the pyridine ring of molecule $B[\mathrm{C} 19 \cdots C g 2=3.868$ (3) $\AA$, $\mathrm{H} 19 B \cdots C g 2=3.37 \AA$ and $\mathrm{C} 19-\mathrm{H} 19 B \cdots C g 2=114^{\circ} ; C g 2$ is the centroid of the $\mathrm{C} 1^{\prime}-\mathrm{C} 5^{\prime} / \mathrm{N} 1^{\prime}$ ring].

There are also two intermolecular $\mathrm{C}-\mathrm{H} \cdots \pi$ interactions (Desiraju \& Steiner, 1999) linking symmetry-related molecules $A$ and $B$, and these are dominant in the structure of (I), providing strong packing. In the first of these interactions, phenyl atom C10 of molecule $A$ interacts with the symmetryrelated phenyl ring of molecule $B$ at $\left(1+x, \frac{3}{2}-y, \frac{1}{2}+z\right)$ $[\mathrm{C} 10 \cdots \mathrm{Cg} 3=3.498$ (4) $\AA, \mathrm{H} 10 \cdots C g 3=2.71 \AA$ and $\mathrm{C} 10-$
$\mathrm{H} 10 \cdots C g 3=143^{\circ} ; C g 3$ is the centroid of the $\mathrm{C1}^{\prime}-\mathrm{C}^{\prime} / \mathrm{N} 1^{\prime}$ ring at $\left.\left(1+x, \frac{3}{2}-y, \frac{1}{2}+z\right)\right]$. In the second interaction, orthosubstituted phenyl atom $\mathrm{C} 17^{\prime}$ of molecule $B$ interacts with the symmetry-related phenyl ring of molecule $A$ at $(x-1, y, z)$ $\left[\mathrm{C} 17^{\prime} \cdots \mathrm{Cg} 4=3.653(4) \AA, \mathrm{H} 17^{\prime} \cdots \mathrm{Cg} 4=2.95 \AA\right.$ and $\mathrm{C}^{\prime} 7^{\prime}-$ $\mathrm{H} 17^{\prime} \cdots \mathrm{Cg} 4=134^{\circ} ; C g 4$ is the centroid of the $\mathrm{C} 1-\mathrm{C} 5 / \mathrm{N} 1$ ring at $(x-1, y, z)]$.

## Experimental

The title compound was synthesized according to the method of Shah (2000); full details of the synthesis will be published elsewhere. Thin plate-like crystals of (I) were grown by slow evaporation from a solution in a mixture of chloroform and ethanol (1:1).

## Crystal data

$\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}$
$M_{r}=300.31$
Monoclinic, $P 2_{1} / c$
$a=8.5542$ (10) A
$b=32.027$ (3) $\AA$
$c=12.3910$ (7) $\AA$
$\beta=104.493$ (3) ${ }^{\circ}$
$V=3286.6(5) \AA^{3}$
$Z=8$
$D_{x}=1.214 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}=1.204 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}$ measured by flotation in an aqueous potassium iodide solution
$\mathrm{Cu} K \alpha$ radiation
Cell parameters from 25 reflections
$\theta=25-35^{\circ}$
$\mu=0.60 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Plate, colourless
$0.2 \times 0.1 \times 0.1 \mathrm{~mm}$

## Data collection

Enraf-Nonius CAD-4
$R_{\text {int }}=0.061$
$\theta_{\text {max }}=67.9^{\circ}$
$h=-10 \rightarrow 10$
$k=0 \rightarrow 30$
$l=0 \rightarrow 14$
2 standard reflections frequency: 60 min intensity decay: $1 \%$

H -atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0770 P)^{2}\right.$
$+0.3042 P$ ]
$(\Delta / \sigma)_{\max }=0.050$
$\Delta \rho_{\text {max }}=0.13 \mathrm{e}_{\mathrm{m}} \AA^{-3}$
$\Delta \rho_{\text {min }}=-0.21 \mathrm{e} \mathrm{A}^{-3}$

5612 independent reflection
3150 reflections with $I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.051$
$w R\left(F^{2}\right)=0.160$
$S=1.02$
5612 reflections
426 parameters
diffractometer
$\omega / 2 \theta$ scans
Absorption correction: empirical
via $\psi$ scan (North et al., 1968)
$T_{\text {min }}=0.930, T_{\text {max }}=0.940$
6001 measured reflections

Table 1

Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| O1-C1 | 1.336 (3) | $\mathrm{Ol}^{\prime}-\mathrm{C} 19^{\prime}$ | 1.431 (4) |
| :---: | :---: | :---: | :---: |
| O1-C19 | 1.463 (3) | $\mathrm{N} 1^{\prime}-\mathrm{Cl}^{\prime}$ | 1.299 (3) |
| N1-C1 | 1.313 (3) | $\mathrm{N} 2^{\prime}-\mathrm{C} 18^{\prime}$ | 1.133 (4) |
| N2-C18 | 1.142 (3) | $\mathrm{C} 2^{\prime}-\mathrm{C} 18^{\prime}$ | 1.444 (4) |
| C2-C18 | 1.437 (3) | C19'- ${ }^{\text {c } 20}{ }^{\prime}$ | 1.337 (6) |
| C19-C20 | 1.477 (4) | $\mathrm{C} 19^{\prime}-\mathrm{C} 20 \mathrm{~A}$ | 1.394 (8) |
| $\mathrm{O} 1^{\prime}-\mathrm{C} 1^{\prime}$ | 1.350 (3) |  |  |
| C1-O1-C19 | 117.81 (19) | $\mathrm{C1}^{\prime}-\mathrm{Ol}^{\prime}-\mathrm{C} 9^{\prime}$ | 119.0 (3) |
| $\mathrm{C} 1-\mathrm{N} 1-\mathrm{C} 5$ | 118.1 (2) | $\mathrm{C} 1^{\prime}-\mathrm{N} 1^{\prime}-\mathrm{C} 5^{\prime}$ | 117.6 (2) |
| N1-C1-C2 | 123.9 (2) | $\mathrm{N} 1^{\prime}-\mathrm{C1}^{\prime}-\mathrm{C}^{\prime}{ }^{\prime}$ | 124.6 (3) |
| N2-C18-C2 | 177.2 (3) | $\mathrm{N} 2^{\prime}-\mathrm{C} 18^{\prime}-\mathrm{C} 2^{\prime}$ | 179.5 (3) |
| O1-C19-C20 | 107.6 (2) | $\mathrm{C} 20^{\prime}-\mathrm{C} 19^{\prime}-\mathrm{O}^{\prime}$ | 110.3 (4) |
| N1-C5-C12-C17 | 154.8 (2) | $\mathrm{N} 1^{\prime}-\mathrm{C}^{\prime}-\mathrm{C} 12^{\prime}-\mathrm{C17}^{\prime}$ | 156.8 (3) |
| C4-C3-C6-C11 | -142.1 (2) | $\mathrm{C} 2^{\prime}-\mathrm{C} 3^{\prime}-\mathrm{C}^{\prime}-\mathrm{Cl1}^{\prime}$ | -41.7 (4) |

Figure 2
The molecular packing in (I), viewed along the $b c$ plane, showing the hydrogen-bonding interactions.


Table 2
Hydrogen-bonding geometry $\left(\AA{ }^{\circ},{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{C}^{\prime} 9^{\prime}-\mathrm{H} 19 D \cdots \mathrm{~N} 1^{\prime}$ | 0.97 | 2.34 | $2.709(5)$ | 102 |
| $\mathrm{C}^{\prime} 5^{\prime}-\mathrm{H} 15^{\prime} \cdots \mathrm{N} 2^{\mathrm{i}}$ | 0.93 | 2.55 | $3.399(4)$ | 153 |

Symmetry code: (i) $x, \frac{3}{2}-y, z-\frac{1}{2}$.
H atoms were treated as riding, with $\mathrm{C}-\mathrm{H}$ distances in the range 0.93-0.97 Å.

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: MolEN (Fair, 1990); data reduction: MolEN; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP (Johnson, 1965) and PLATON (Spek, 1997); software used to prepare material for publication: SHELXL97.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: DE1195). Services for accessing these data are described at the back of the journal.

## References

Black, S. N., Davey, R. J., Slawin, A. M. Z. \& Williams, D. J. (1992). Acta Cryst. C48, 323-325.
Bolte, M. (1998). Acta Cryst. C54, 132-134.
Desiraju, G. R. \& Steiner, T. (1999). The Weak Hydrogen Bond in Structural Chemistry and Biology. New York: Oxford University Press Inc.
Enraf-Nonius (1989). CAD-4 Software. Version 5.0. Enraf-Nonius, Delft, The Netherlands.
Fair, C. K. (1990). MolEN. Enraf-Nonius, Delft, The Netherlands.
Fonseca, I., Martínez-Carrera, S. \& García-Blanco, S. (1986). Acta Cryst. C42, 1792-1794.
Hussain, Z., Fleming, F. F., Norman, R. E. \& Chang, S.-C. (1996). Acta Cryst. C52, 1010-1012.
Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
Lokaj, J., Vrábel, V., Sivý, P., Kettmann, V., Ilavský, D. \& Ječný, J. (1991). Acta Cryst. C47, 886-888.
North, A. C. T., Phillips, D. C. \& Mathews, F. S. (1968). Acta Cryst. A24, $351-$ 359.

Patel, U. H., Dave, C. G., Jotani, M. M. \& Shah, H. C. (2002a). Acta Cryst. C58, o191-o192.
Patel, U. H., Dave, C. G., Jotani, M. M. \& Shah, H. C. (2002b). Z. Kristallogr. New Cryst. Struct. 217, 29-31.
Patel, U. H., Dave, C. G., Jotani, M. M. \& Shah, H. C. (2002c). Z. Kristallogr. New Cryst. Struct. 217, 32-34.
Perez-Medina, L. A., Merriella, R. P. \& McElvain, S. M. (1947). J. Am. Chem. Soc. 69, 2574-2579.
Shah, H. C. (2000). PhD Thesis, Gujarat University, India.
Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
Spek, A. L. (1997). PLATON. University of Utrecht, The Netherlands.


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